# **Experimenting with a Chemical Change**

**Experiment Activity** 

### **Materials** (per group)

measuring cup
3 plastic cups, 10 oz
3 antacid tablets
resealable plastic bag
grid paper

safety goggles water (room temperature) clock with second hand metal spoon

#### **Material Substitutions**

Effervescent candy tablets or bath tablets may be used in place of antacid tablets.

#### **Advance Prep**

Allow the water to sit out for several hours, until it is room temperature.

#### **Hints and Tips**

- Be sure to explain that to "completely break down" means that the tablet is completely gone, and there is no more "fizzing." There may still be some residue at the bottom of the cup.
- When using the tablet broken into four pieces, the pieces do not have to be exactly the same size.

#### **Safety Note**

- Remind students to clean up all spills immediately
- Remind students not to drink the liquid or eat any of the antacid.

#### **Additional Comments**

You may wish to combine the data from all groups for more accurate data collection. Any group with data that is very different from the class data may wish to repeat the experiment. This would be a good opportunity to discuss the necessity of repeating experiments to obtain more reliable results.

Activity Rubrics Scoring	g Ke	<b>Э</b> У	4	corre deta	comp	lete,		partia comp			parti com	ially plete	corre e, lac	ect, p ks so	artia ome (	lly detai				or in sistar	olete,
<b>Experiment Activity</b> Experimenting with a Chemical Change																					
Scoring Criteria	V /		/ /	/ /		/ /	' /		/ /	/ /			/ /					/ ,	/ ,	/ /	
Student made a hypothesis about an antacid tablet breaking down in water.																					
Student controlled variables and followed instructions to experiment with antacid tablets.																					
Student measured the time it took for the tablets to break down.																					
Student collected and interpreted data by making a chart and a bar graph.																					
Student communicated by stating conclusion.																					
Score total points																					
% equivalent																					7

Date



# **Experimenting with a Chemical Change**

## **State the Problem**

How does changing the shape of an antacid tablet affect how fast it breaks down in water?

# **Formulate Your Hypothesis**

If you break apart or crush an antacid tablet, will it break down in water faster than, slower than, or at the same rate as a whole tablet? Write your hypothesis.

# **Identify and Control the Variables**

The shape of the tablet is the variable you can change.

# **Test Your Hypothesis**

**1–6.** Follow the steps on textbook pages B25–B26 to perform your experiment.

## **Collect Your Data**

Record your data in the chart.

Trial	Tablet shape	Time to break down
1	Whole	
2	4 pieces	
3	Crushed	

# **Interpret Your Data**

Use the data from your chart to make a bar graph.

### **Breakdown of Tablet**

	60			
Time for	50			
tablet to	40			
break	30			
down	20			
(seconds)	10			
	0 1	Whole	4 pieces	Crushed
			Shape of tablet	

Name Date	
Describe any patterns you see in the time it took for the table down.	ts to break
State Your Conclusion	
How do your results compare with your hypothesis? Commun conclusion. Explain how changing the shape of the tablet affetime it took to break down.	_
Inquire Further  If you increase the temperature of the water, will a tablet break faster? Can you think of other ways to make the tablet break faster? Develop a plan to answer these or other questions you	down
Self-Assessment Checklist	
I made a <b>hypothesis</b> about an antacid tablet breaking down in water.	
I <b>identified</b> and <b>controlled variables</b> , and I followed instructions to perform an <b>experiment</b> with antacid tablets.	
I measured the time it took for the tablets to break down.	
I <b>collected</b> and <b>interpreted</b> data by making a chart and a bar graph.	
I <b>communicated</b> by stating my conclusion.	



**Notes for Home** Your child conducted an **experiment** to determine how the shape of an antacid tablet affects how fast it breaks down in water.

**Home Activity:** Ask your child to explain why a 5-pound block of ice would last longer in an ice chest than 5 pounds of crushed ice.

Date



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